

Structure of Matter – MidTerm3

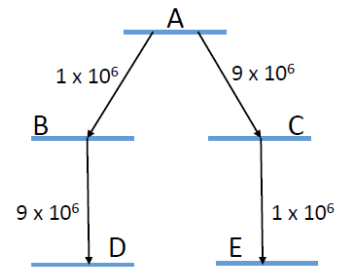
May 07, 2020

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Motivate your answers!

PROBLEM 1. [13 pnts]

- a) Calculate the proton density [per fm³] of a ${}^{261}_{104}\text{Rf}$ nucleus. [2 pnts]
- b) The Rutherfordium isotope ${}^{261}_{104}\text{Rf}$ can decay to nobelium (No) via α decay or to lawrencium (Lr) via β^+ decay. Give the complete formula for each of the decay reactions. (mind charge conservation) [3 pnts]
- c) As is shown in the figure a nucleus A can decay via two channels creating two granddaughter nuclei D and E. Possible decay pathways are ABD and ACE. The transition rates per second are indicated in the figure. If initially 100,000 nuclei A were produced in an experiment how many granddaughters D and E are finally produced? [2 pnts].
- d) Are both pathways equally fast? [2 pnts].
- e) Use the liquid drop model approximation to verify whether ${}^{64}_{30}\text{Zn}$ can or cannot decay to ${}^{64}_{29}\text{Cu}$. [4 pnts]



$$B(N, Z) = aA - bA^{\frac{2}{3}} - \frac{dZ^2}{A^3} - s \frac{(N-Z)^2}{A} - \frac{\delta}{A^{\frac{1}{2}}}$$

a=	15.8	$\delta=$	11.2	odd-odd
b=	18.3		0	even-odd
d=	0.71		-11.2	even-even
s=	23.2		all parameters are in MeV	

PROBLEM 2. [12 pnts]

Use the generic sequence of nuclear shell filling: 1s, 1p, 1d, 2s, 1f, 2p, 1g,

- a) Which of the tabulated nuclei have nuclear spin $J=7/2$? [3 pnts]
- b) Which of the tabulated nuclei have even parity? [2 pnts]
- c) Hyperfine structure: consider a Sc isotope with $I=4$. Sc has a ${}^2D_{3/2}$ level as its ground electronic level. What are the possible hyperfine levels F ? [2 pnts]
- d) Show that the space (number of states) spanned by I and J is the same as in the hyperfine description F . [2 pnts].
- e) Back to the nucleus: Take $A < 45$, what is the lightest and heaviest isotope of Sc, that in principle could have a nuclear spin of 4? [3 pnts]

element
${}^{45}_{21}\text{Sc}$
${}^{48}_{22}\text{Ti}$
${}^{51}_{23}\text{V}$
${}^{52}_{24}\text{Cr}$
${}^{55}_{25}\text{Mn}$
${}^{56}_{26}\text{Fe}$
${}^{59}_{27}\text{Co}$
${}^{59}_{28}\text{Ni}$
${}^{63}_{29}\text{Cu}$